

# Nonequilibrium entropy and the second law of thermodynamics

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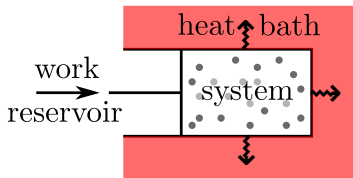
Vienna, May 2020

*“Entropy.” — “Entropy?” — “Yeah, entropy. Boris explained it.  
It’s why you can’t get the toothpaste back in the tube.”*

## Outline

- nonequilibrium thermodynamics: phenomenology
- thermodynamic entropy out of equilibrium
- entropy production as change in thermodynamic entropy

## The Nonequilibrium Second Law (Clausius, 1865)



### Today

For ease of presentation only **one heat bath**.

### the concept of entropy production:

( $S =$  *thermodynamic* entropy!)

$$\begin{aligned} 0 &\leq \Delta S_{\text{universe}} \\ &= \Delta S_S + \Delta S_{\text{env}} \quad (\text{weak coupling}) \\ &= \Delta S_S - \int \frac{dQ}{T} \quad (\text{ideal bath}) \\ &= \Delta S_S - \frac{Q}{T_0} \quad (\text{weakly perturbed bath, } T_0 = \text{initial temp.}) \end{aligned}$$

## Quantum-Classical Dictionary

	quantum	classical
$\mathcal{H}$	Hilbert space	phase space $\Gamma$
$\rho$	density matrix	phase space distribution
$H$	Hamiltonian	Hamiltonian
$\partial_t \rho$	$= -\frac{i}{\hbar}[H, \rho]$	$= \{H, \rho\}$
$\text{tr}\{\dots\}$	trace operation	phase space integral $\int d\Gamma \dots$
$S_{\text{vN}}(\rho)$	$= -\text{tr}\{\rho \ln \rho\}$	$= -\int d\Gamma \rho(\Gamma) \ln[\rho(\Gamma)/h^{Nd}]$
$\Pi_x$	projector with outcome $x$	characteristic function for a set $x$ : $\Pi_x(\Gamma) = 1$ if $\Gamma \in x$ , otherwise 0
$V_x$	rank $\text{tr}\{\Pi_x\}$	volume $\int d\Gamma \Pi_x(\Gamma)/h^{Nd}$

(today  $k_B \equiv 1$ )

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## Proposal 1

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Problem

$$\frac{d}{dt} S_{\text{vN}}[\rho(t)] = 0.$$

## Proposal 2

$$S = S_{\text{Boltzmann}}(x) = \ln V_x$$

$x$  – some macroscopic constraint(s)



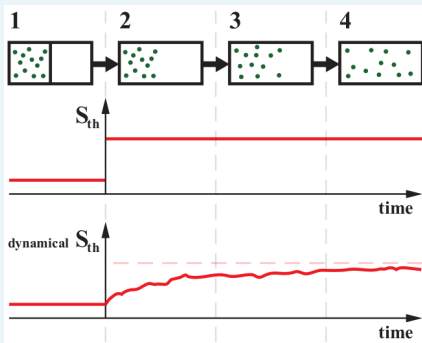
## Proposal 2

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Problem (Šafránek, Deutsch & Aguirre, arXiv 1905.03841)

- little dynamical information
- typically zero for small systems ( $V_x = \text{tr}\{\Pi_x\} = 1$ )



## Proposal 3

### Observational entropy

$$S = S_{\text{obs}}^X(\rho) = \sum_x p_x (-\ln p_x + \ln V_x)$$

with  $X = \sum_x x \Pi_x$  some (typically coarse-grained) observable,  
 $p_x = \text{tr}\{\Pi_x \rho\}$ , and  $V_x = \text{tr}\{\Pi_x\}$ .

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#### History

- first introduced by von Neumann (or Wigner?) in Z. Phys. **57**, 30 (1929)
- closely related: “coarse-grained entropy” for classical systems see, e.g., the Ehrenfests, *Begriffliche Grundlagen der statistischen Auffassung in der Mechanik* (1911), Wehrl, Rev. Mod. Phys. **50**, 221 (1978); goes back to an idea of Gibbs.
- recently revived by Šafránek, Deutsch & Aguirre, Phys. Rev. A **99**, 010101 (2019)

## Properties of Observational Entropy (1/2)

Šafránek, Deutsch & Aguirre, PRA 99, 012103 (2019)

### (1) boundedness

$$S_{\text{vN}}(\rho) \leq S_{\text{obs}}^X(\rho) \leq \ln \dim \mathcal{H}$$

### (2) extensivity

If  $X = X_1 \otimes \cdots \otimes X_n$  and  $\rho = \rho_1 \otimes \cdots \otimes \rho_n$ , then

$$S_{\text{obs}}^X(\rho) = \sum_{i=1}^n S_{\text{obs}}^{X_i}(\rho_i).$$

### (3) equivalence

$$S_{\text{obs}}^X(\rho) = S_{\text{vN}}(\rho) \quad \Leftrightarrow \quad \rho = \sum_x p_x \omega(x) \quad (p_x \text{ arbitrary})$$

with  $\omega(x) \equiv \Pi_x / V_x$  (generalized microcanonical ensemble).

## Properties of Observational Entropy (2/2)

Strasberg & Winter, arXiv 2002.08817

Now consider dynamics

*Change of  $S_{\text{obs}}^{X_t}[\rho(t)]$  with  $X = X_t$ , initial time  $t = 0$ .*

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(4) “second law” (see also Gibbs, Lorentz, Wehrl, Zubarev, ...)

If  $S_{\text{obs}}^{X_0}[\rho(0)] = S_{\text{vN}}[\rho(0)]$ , then  $\Delta S_{\text{obs}}^{X_t}(t) \geq 0$ .

(5) fluctuation thm. see also Schmidt, Gemmer, Z Naturforsch A 75 265, 2020

If  $S_{\text{obs}}^{X_0}[\rho(0)] = S_{\text{vN}}[\rho(0)]$ , then

$$\left\langle e^{-\Delta s_{\text{obs}}^{X_t}(t)} \right\rangle = \sum_{x_t, x_0} p(x_t, x_0) e^{-\Delta s_{\text{obs}}^{X_t}(t)} = 1,$$

where

$$p(x_t, x_0) = \text{tr}\{\Pi_{x_t} U(t) \Pi_{x_0} \rho(0) \Pi_{x_0} U(t)^\dagger\},$$
$$\Delta s_{\text{obs}}^{X_t}(t) = s_{\text{obs}}^{X_t}(t) - s_{\text{obs}}^{X_0}(0) = -\ln \frac{p_{x_t}}{V_{x_t}} + \ln \frac{p_{x_0}}{V_{x_0}}.$$

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## Energy measurements

spectral decomposition:  $H|E_i\rangle = E_i|E_i\rangle$

coarse-grained projector:  $\Pi_E = \Pi_{E,\delta} = \sum_{E_i \in [E, E+\delta)} |E_i\rangle\langle E_i|$

measured observable:  $X = \sum_E E \Pi_E \xrightarrow{\delta \rightarrow 0} H$

observational entropy:  $S_{\text{obs}}^E(\rho) = \sum_E p_E (-\ln p_E + \ln V_E)$

equilibrium states:  $\Omega = \left\{ \sum_E p_E \omega(E) \mid p_E \text{ arbitrary} \right\}$

(note:  $\delta$  is a free parameter in theory, assumed to be chosen small enough, left implicit in the notation)



## Heat and work in isolated systems

$$\partial_t \rho(t) = -\frac{i}{\hbar} [H(\lambda_t), \rho(t)], \quad \rho(0) \in \Omega(\lambda_0)$$

standard definition of work

$$\Delta U(t) = \text{tr}\{H(\lambda_t)\rho(t) - H(\lambda_0)\rho(0)\} \equiv W_{\text{tot}}(t) (\geq 0)$$

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$$W_{\text{rec}}(t) \equiv \Delta U(t) - \int_0^t T_s^* dS_{\text{obs}}^E \left[ \frac{e^{-H(\lambda_s)/T_s^*}}{Z(\lambda_s)} \right]$$

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### Theorem (reversible process)

$$W_{\text{rec}}(t) = W_{\text{tot}}(t) \quad \Leftrightarrow \quad \Delta S_{\text{obs}}^E(t) = 0 \quad \Leftrightarrow \quad \rho(t) \in \Omega(\lambda_t)$$

$$\text{otherwise } Q_{\text{rem}}(t) \equiv \Delta U(t) - W_{\text{rec}}(t) > 0$$

## Entropy production in open systems

$$H_{SB}(\lambda_t) = H_S(\lambda_t) + V_{SB} + H_B, \rho_{SB}(0) \in \rho_S(0) \otimes \Omega_B$$

- chosen observable:  $X_t = S_t \otimes E_B$

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$$\Delta S_{\text{obs}}^{S_t}(t) + \int_0^t \frac{\dot{Q}_{\text{rem}}^B(s)}{T_s^*} ds \geq \Delta S_{\text{obs}}^{S_t, E_B}(t) \geq 0$$

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- depending on the level of control of  $H_S(\lambda_t)$ :

$$Q_{\text{rem}}(t) = \left\{ \begin{array}{l} W_{\text{tot}} - \Delta F_S^{\text{eq}} \\ W_{\text{tot}} - \Delta F_S^{\text{noneq}} \end{array} \right\} \geq 0.$$

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## Summary (Strasberg & Winter, arXiv 2002.08817)

see also Strasberg, arXiv 1906.09933

Observational entropy...

- ...seems to be a good candidate for thermodynamic entropy out of equilibrium (compare also with Šafránek, Deutsch & Aguirre)
- ...provides a *consistent* derivation of the second law for open systems (“the thermodynamic entropy of the universe never decreases!”)
- ...provides an extremely flexible tool (one or multiple bath, beyond Gibbs states, including a large class of correlated states, etc.)

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