

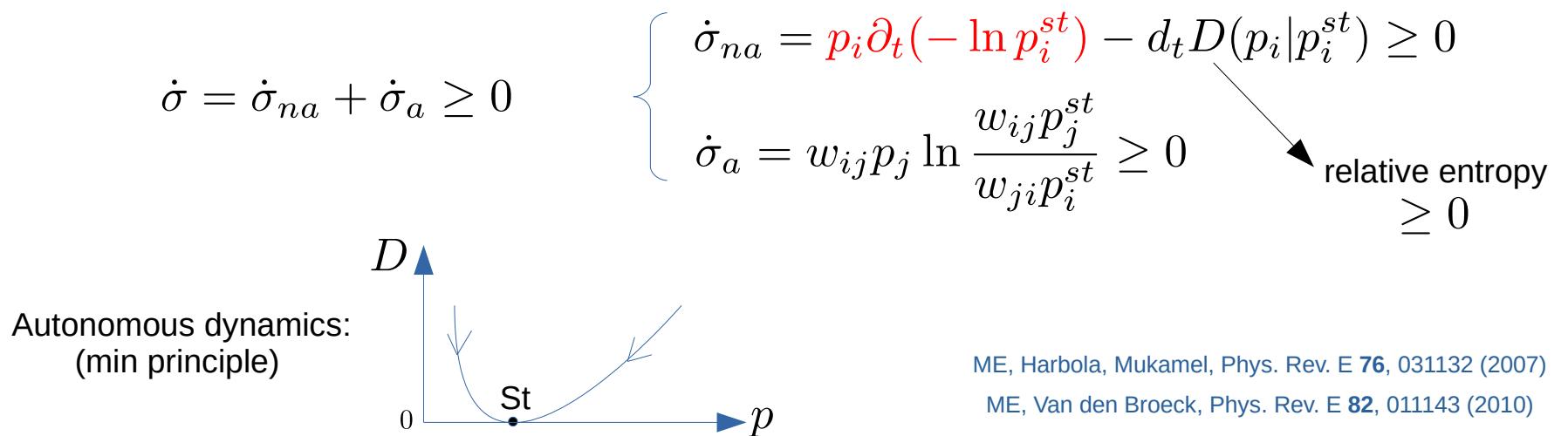
Entropy and MaxEnt in NonEq Thermodynamics

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Markovian master equation (non-autonomous): $d_t p_i = w_{ij} p_j$

Mathematical “Entropy Production”: $\dot{\sigma} = w_{ij} p_j \ln \frac{w_{ij} p_j}{w_{ji} p_i} \geq 0$

Instantaneous steady state: $0 = w_{ij} p_i^{st}$ “Equilibrium”: $w_{ij} p_j^{eq} = w_{ji} p_i^{eq} \quad \forall i, j \rightarrow \dot{\sigma} = 0$



Physics

Thermodynamics: system exchanging conserved quantities with reservoirs with conjugated intensive variables

Local detailed balance: $\ln \frac{w_{ij}}{w_{ji}}$ is an entropy change in the reservoirs (and possibly internal)

Physical Entropy Production: $\dot{\sigma} = \partial_t(\Phi^{eq} - \Phi) - d_t D(p|p^{eq}) + A_\alpha J_\alpha \geq 0$

Rao, ME, New J. Phys. **20**, 023007 (2018)

Rao, ME, Entropy **20**, 635 (2018)

$$\Phi^{eq} - \Phi \geq 0 \quad \text{fluxes-forces}$$

Massieu potential e.g. $\Phi = S - \frac{1}{T_0}E + \frac{\mu_0}{T_0}N \dots$

Autonomous detailed balance dynamics $\rightarrow \Phi$ is maximized

$$A_\alpha = 0 \quad \forall \alpha$$

"Jaynes's maxent is realized dynamically"

In general no thermodynamic max or min principle!

Close to equilibrium (minimum EP principle): $d_t \dot{\sigma} = d_t \dot{\sigma}_{na} \leq 0$

